



Calculation of structural properties and vibrational frequencies of NiO and Ni_2O_3 crystals

7 J]U]cb. H\Y'>ci fbU'cZ7\Ya JW'D\ng]Mg'6Zz' %\$'f% +) t/Xc]. '%"\$%\$*' #%(' \$, , ,
J]Yk 'cb]bY. \hd.#Xl "Xc]cf[#%"%\$*' #%(' \$, , ,
J]Yk 'HUV'cZ7 cbYbfg. \hd.#gV]U]cb"U]d]cf[#VcbYbH]U]d]#ci fbU'#Vd]# &# 3j Yf1dXZ]tj
Di V]g\YX VmiH Y 5D'Di V]g\]b]

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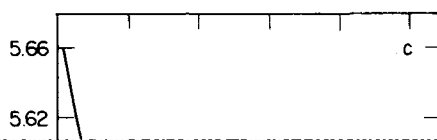
: Uf]bZUFYX'gdYVfcgVtdmiUbX'Ya d]f]W]]bYfa c'YW'Uf'dcYbH]U'Zcf' 'UbX' B&i bXYf'dfYggi fY'
>"7\Ya "'D\ng"'74Z'&* +) 'f% , %&/'%\$"%\$*' #%(('%(' *'

7 ca a Ybhc] J]Vfcb'UbX'UH]W'Z'YeI YbVng\]Zg]]b'h.Y'F Ua Ub'gdYVfU'cZgc'X' B&'UbX' B&'UbX']VfU]cbU'
ZcfW'Vtbg]U]bfg'cZ]U]ca]Wa c'YW'Uf'Wng]U'g'
>"7\Ya "'D\ng"'71Z' , - ' f% +- t/'%\$"%\$*' #%(' , , \$&'

J]Vfcb'UbX'UH]W'Z'YeI YbVng\]Zg]]b'h.Y'F Ua Ub'gdYVfU'cZgc'X' B&'UbX' B&'UbX']VfU]cbU'ZcfW'Vtbg]U]bfg'cZ'
X]U]ca]Wa c'YW'Uf'Wng]U'g'
>"7\Ya "'D\ng"'67Z' *) - f% ++t/'%\$"%\$*' #%(')' \$*'

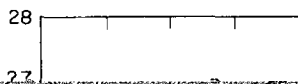
F Ua Ub'gdYVfU'cZgc'X' B&'UbX'
>"7\Ya "'D\ng"'59Z'*) f% +' t/'%\$"%\$*' #%"%\$* +-- , -'

F Ua Ub'gdYVfU'cZgc'X' B&'UbX' B&i bXYf\]]'dfYggi fY'Uh("&?'
>"7\Ya "'D\ng"'59Z'()) - f% +' t/'%\$"%\$*' #%"%\$* , \$*)'



$$K = 3244.18 \text{ kcal/\AA}^2$$

for the harmonic intramolecular potential [Eq. (19)].
The molecular bond length b_0 was not treated as a pa-
rameter but set at the experimental N-N bond length of



VI) in the volume range of 40–55 Å³/molecule. At this range the graph of $d \ln \omega_i$ vs $d \ln V$ was practically linear

