





example of $\text{Fe}^{2+}/\text{Co}^{2+}$ hexagonal ground state ordered structures [18]. For

example, the $\text{Fe}^{2+}/\text{Co}^{2+}$ hexagonal ground state ordered structures [18] are characterized by the transfer, polarization,

relaxation of the C_subl , and the effect of the C_subl on the spin transfer, polarization,

The components of the local cell parameter is given by $a_\text{cell}^{(n)} = \frac{1}{L} \sum_{i=1}^L a_i^{(n)}$, where $a_i^{(n)}$ is the equilibrium lattice cell length of the i -th cell, and we can express the mixing enthalpy of the two species of cells as

$$\Delta H_\text{mix}(x_\text{c,T}) = \frac{c}{L} \sum_{i=1}^L \left(\frac{a_i^{(n)}}{a_\text{cell}^{(n)}} - 1 \right)^2 + G(x_\text{c,T}), \quad (3)$$

$x_\text{c,T}$ is the volume fraction of the mixture of AC and BC to $x_\text{c,T} = 0.5$. The two terms of (3) reflect the dual coordinates used in phenomenological models of solid solutions [19], i.e., the spatial variability [17, 31], and mixing enthalpies [3], e.g., the Darken-Gurney [2] and Floudas-Jilland models.

Many approximations have been previously adopted in the literature to calculate the mixing enthalpy of the two species of cells.

For example, the local cell length of each cell is assumed to be constant [18, 32–34],



Fig. 1. The vector coarse approach.

$$M = M_\text{II} \cup M_\text{W} \cup M_\text{M},$$

$$M_\text{II} = M_\text{II}^\text{I} \cup M_\text{II}^\text{II} \cup M_\text{II}^\text{III},$$

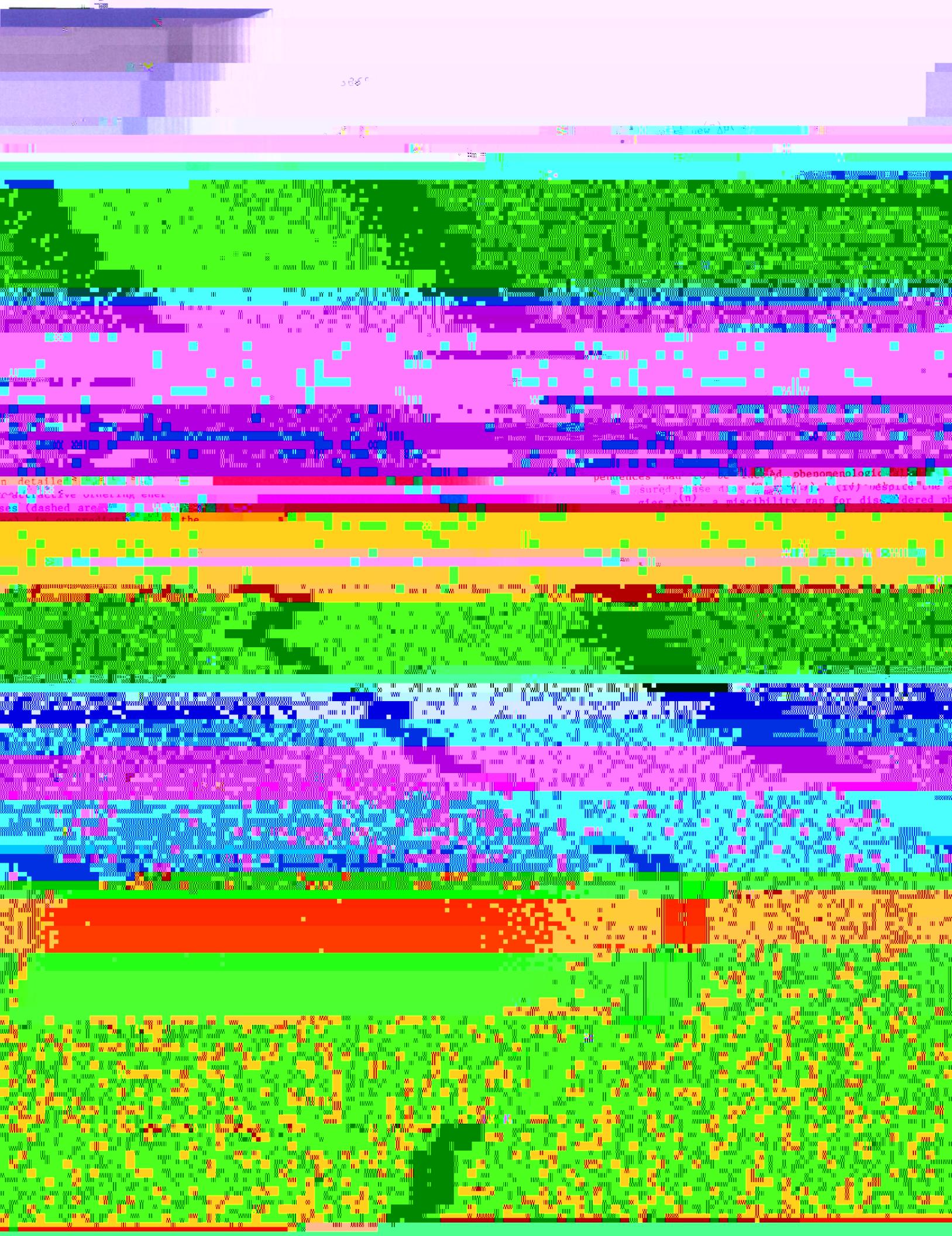
$$M_\text{W} = M_\text{W}^\text{I} \cup M_\text{W}^\text{II} \cup M_\text{W}^\text{III},$$

$$M_\text{M} = M_\text{M}^\text{I} \cup M_\text{M}^\text{II} \cup M_\text{M}^\text{III}.$$

lattice parameter a_{fkl} and

K. In the case of V-P-Co alloy





t the new $\Delta H^{(n)}$ values produce ---
without any adjustment¹⁸ of
that the elastic energies are
Furthermore, $\Delta E^{(n)}(a)$ can
be zero in some cases.

Table 1: Components of Eq. (3) for $\text{GaN}_{4-n}\text{P}_n$ compounds in N_2O_2 pairs, lattice constants in Å.

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