Theoretical and experimental studies of the ZnSe/CulnSe₂ heterojunction band offset

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(Received 5 January 1993; accepted for publication 18 March 1993)

We report first-principles band structure calculations that show that $ZnSe/CuInSe_2$ has a significant valence band offset (VBO, ΔE_v): 0.70 ± 0.05 eV for the relaxed interface and 0.60 ± 0.05 eV for the coherent interface. These large values demonstrate the failure of the common anion rule. This is traced to a stronger Cu,d-Se,p level repulsion in $CuInSe_2$ than the Zn,d-Se,p repulsion in ZnSe. The VBO was then studied by synchrotron radiation soft x-ray photoemission spectroscopy. ZnSe overlayers were sequentially grown in steps on n-type $CuInSe_2(112)$ single crystals at 200 °C. In situ photoemission measurements were acquired after each growth in order to observe changes in the valence band electronic structure as well as changes in the In 4d and Zn 3d core lines. Results of these measurements reveal that the VBO is $\Delta E_n = 0.70\pm0.15$ eV, in good agreement with the first-principles prediction.

The ternary $A^{\rm I}B^{\rm III}X_2^{\rm VI}$ chalcopyrite semiconductor, CuInSe₂ (E_g =1.1 eV), has received considerable attention as an absorber in heterojunction solar cells. Technical ad-

14%. The theoretical electronic structure of CuInSe₂ has been calculated and experimental data have confirmed the

continuity and the results of a soft x-ray photoemission investigation of heterojunction formation.

We have first calculated the valence band offset (VBO) in 7nSe/CuInSe, then measured it Pravious theories of

junction partner is a ternary compound.

The band offset between two materials ABX_2 and DX

applications as a window layer for photovoltaic heterojunctions.⁴

The $ZnSe/CuInSe_2$ heterojunction provides an inter-

where $\Delta E_{vbm,c}^{ABX_2} = E_{vbm}^{ABX_2} - E_c^{ABX_2}$ and $\Delta E_{vbm,c}^{DX} = E_{vbm}^{DX} - E_c^{DX}$ are the core-level (c) to the valence band maximum (vbm) energy separations for materials ABX_c and DX respec-

ishing valence band offset. We now know⁶ that this rule is violated in the above two cases involving *binary* heterojunction partners. ZnSe/CuInSe₂ is the first nearly lattice

heterojunction partner to be tested in this respect. This system is interesting because it involves a near-transition metal atom (copper): the failure of the common-anion rule in Zn, Cd, and Hg containing II-VI heterojunctions was attributed⁶ to the different repulsion between the cation d orbitals and the common anion p orbitals on either side of the interfece. Since the d band in Zn, Cd, and Hg-based II-VIs,² there is a correspondingly large difference in the p-d level repulsions in CuInSe₂ and II-VI materials. Hence, Cu-containing heterojunctions with II-VI semiconductors should exhibit a greater departure from the common-anion rule than pure II-VI heterojunctions. Progress in attaining

continuities is necessary in order to ultimately control these interfacial properties. This letter discusses theoretical

CuInSe₂ heterojunction in an analogous way to previous calculations on binary systems.⁶ We have performed first-principles total energy and band structure calculations for the constituents CuInSe₂ and ZnSe, obtaining the three

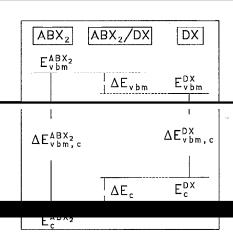
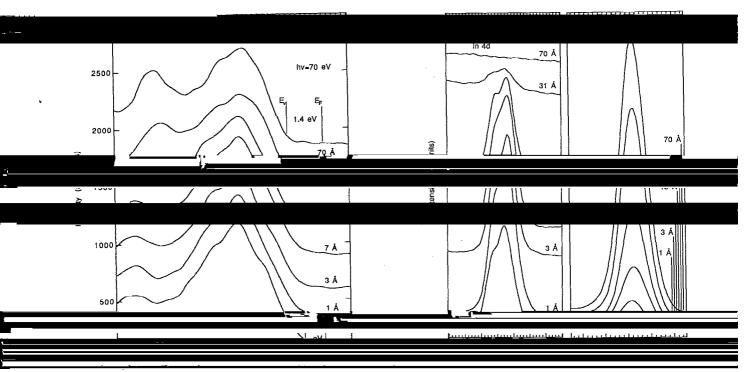


FIG. 1. Schematic energy-level diagram used to deduce the valence band

tions of the 7-De /Culoc



Binding Energy (eV)

FIG. 2. Normal emission valence band spectra of the $ZnSe/CuInSe_2$ interface as a function of the effective ZnSe coverage for the n-type crystals. Zero of energy is at E_F .

terms of Eq. (1). The two core-to-vbm energy differences $\Delta E_{vbm,c}$ were obtained as eigenvalue differences in separate calculations on CuInSe₂ and ZnSe. The core-level difference ΔE_{c} between the two materials was obtained from

tices with (001) orientation. In this superlattice calculation the number of layers p was increased until the core energy levels of the innermost layer on each side of the interface were bulklike. We found that for p=2 the uncertainty exercisted with using different core levels (e.g., the

FIG. 3. Characteristic core level emission for CuInSe₂ as a function of the effective ZnSe coverage (a) In 4d and (b) Zn 3d. Zero of energy is at E_F .

relaxed, so each heterojunction partner attains its own equilibrium atomic geometry. This gives $\Delta E_{vbm} = 0.70 \pm 0.05$ eV with the *vbm* of CuInSe, above that of ZnSe. In the absence of strain, $\Delta E_{cbm} = E_g^{(1)} - E_g^{(2)} - \Delta E_{vbm} = 0.90 \pm 0.05$ eV, where $E_g^{(1)}$ and $E_g^{(2)}$ are the bulk band gaps of DX and ABX_2 . (ii) ZnSe is assumed to be coherently

field effect; the upper split component moves to lower binding energy by 0.1 eV, hence $\Delta E_{vbm} = 0.60 \pm 0.05$ eV. The tensile strain in ZnSe further lowers its *cbm* by 0.1 eV, so $\Delta E_{cbm} = 0.80 \pm 0.05$ eV. Note that in the presence of strain $\Delta E_{cbm} = 0.80 \pm 0.05$ eV. Note that in the presence of strain

formalism¹¹ as implemented by the general potential, rela-

rule, which states that for a common anion system the

k points for the zinc-blende zone and their equivalent k

structure, the anion p and the cation d orbitals have the same f(F, p) summetry at the zone center, hence they can

calculations. The calculated equilibrium lattice constants for CuInSe₂ and ZnSe are 5.736 Å (observed: 5.784 Å)¹⁵ and 5.618 Å (observed; 5.667 Å),¹⁶ respectively. The calculated tetragonal distortion c/a=1.008 and internal relaxation u=0.214 in CuInSe₂ can be compared with the observed¹⁵ values of 1.004 and 0.224, respectively.

The calculated VBO depends on the interfacial geom-

vbm to lower binding energies by an amount inversely proportional to the (metal-d) to (nonmetal-p) energy difference. In CuInSe₂, the Cu 3d orbitals have a much smaller binding energy than the Zn 3d orbitals in ZnSe, so the (Cu,3d)-(Se,4p) coupling in CuInSe₂ is much stronger than the (Zn,3d)-(Se,4p) coupling in ZnSe. This interaction pushes the vbm of CuInSe₂ up in energy relative to the

as the source of fartare of the common amon fall in 2/1190/

peak positions for clean Culinse₂ is $E_{phm} = \frac{1}{\ln 4d} = \frac{18.55 - E_F}{18.55}$

considerably stronger in the ternary case.

This prediction, obtained prior to experimental testing, was tested by the photoemission measurements described next. The CuInSe₂ crystals were sliced from an ingot which was prepared by high-pressure liquid-encapsulated direc-

sizes. Laue backscattering and x-ray photoemission from a single grain confirmed the (112) surface orientation and stoichiometry. The crystal surface was sputter cleaned with 500 eV Ar ions, 30° incidence followed by annealing for \$\approx 2\$ mm at 500 C to remove sputter-induced damage.

The ZnSe overlayers were deposited in steps on *n*-type CuInSe₂ at 200 °C using a liquid nitrogen shrouded boron nitride effusion cell (750 °C). Photoemission spectra were

opment of the electronic structure at the heterojunction interface. These experiments were performed using the Amoco om torodial grating monochromator (1GM) at the University of Wisconsin Synchrotron Radiation Centering Production and Production Synchrotron Radiation Centering Production of $\Delta E \approx 0.1 \text{ eV}$.

Figure 2 shows the normal emission valence band spectra of the $ZnSe/CuInSe_2$ interface as a function of the effective ZnSe coverage. The observed upper valence band appears as a two peak structure corresponding to the two branches of the $Cu\ d$ bands. The transition of the valence

explication of the electronic structure leading to the 7-6-/

vom for Cuinse₂ and Znse are determined from the linear

erojunction is measured for three coverages (7, 15, and 31 Å); the average over these is 7.72 ± 0.15 eV. From Eq. (1) we have $\Delta E_{vbm} = 17.75-9.25-7.72=0.78\pm0.15$ eV, within the range given by the alternative measurement (Fig. 2) i.e. 0.6 ± 0.1 eV. Bands may bend outside this interfacial

photoemission investigation. The theoretical model predicts an affect of 0.60.0.70 eV depending as the source of failure of the common-anion rule in ZnSe/CuInSe₂. The experimentally determined valence band discontinuity for this heteroignation is 0.7±0.15 eV. Perced on these results

n-type CuInSe₂ can be constructed.

This work was supported by the U.S. Department of Energy under Contract No. DE-AC02-83CH10093. The

under Grant No. DIVIK92-12638.

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is *n*-type, the cleaned $CuInSe_2$ bands are very nearly flat near the surface with minimal (<0.2 eV) band bending. No significant shift is observed in the dominant Cu d band

ZnSe is deposited onto the *n*-type CuInSe₂ crystal surface, the CuInSe₂ valence band remains flat and the structure

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